高圧水素ガスシール用ゴムの内部き裂発生メカニズム についての破壊力学的検討



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1



Introduction-Types of gas leakage-

 ・シール材に求められること ガスがリークしないようにシールすること.
 ・ガスリークの種類は3つに分類される.
 (1)界面リーク
 (2)透過リーク・・・材料のガス透過係数に依存する.
 (3)破壊によるリーク・・・高圧ガス環境下で問題となる.





Introduction-Internal fracture of rubbers by high-pressure gas decompression-



S. Zakaria, and B. J. Briscoe, Chemtech, Vol.20, Aug, pp. 492–495 (1990).





BHR group, "Elastomeric seals for rapid decompression applications in high–pressure service", HSE books (2006).

Introduction-Fracture of O-ring under high-pressure hydrogen gas-



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Introduction-Cavitation by high-pressure gas decompression-

Liquid





Pressurization — Decompression -

Rubber











Introduction-Process of internal fracture due to high-pressure gas decomperssion-

<u>内部破壊プロセス</u>

- ・ゴム分子鎖間から気泡が形成される(液体からの気泡形成と同様).
- ・形成された気泡は分子鎖を破壊し,光学顕微鏡で観察可能なサイズに成長していく. <u>気泡とき裂の定義</u>
- ・気泡:分子鎖間に発生した巨視的な分子鎖切断を伴わないキャビティ
 (光学顕微鏡で観察困難なミクロンサイズ以下)
- ・き裂:気泡を起点として発生する巨視的な分子鎖切断を伴うキャビティ
 - (光学顕微鏡で観察可能なミクロンサイズ以上)
- ・このような内部き裂は,一般的にプリスタと呼ばれる.





Experimental-Material-

- Since cracks initiated from the interior, a transparent rubber material was employed.
- Filler is not added.

Material compound (phr)					
EPDM	100				
Dicumyl peroxide	1.6				
Stearic acid	0.5				
(phr: parts per hundred rubber)					
Density and hardness					
Density (g/cm ³)	0.873				
Hardness	A 51				



Cylindrical specimen (\oplus 29.0 mm \times 12.5 mm)



Experimental-Hydrogen exposure and observation of crack-

- Cylindrical specimen (ϕ 29.0 mm × 12.5 mm) was employed.
- Hydrogen exposure was conducted by using 10 MPa hydrogen vessel at 30 °C.
- After decompression, crack initiation behavior was observed by using optical microscopy in air at room temperature.





Experimental-Hydrogen exposure and observation of crack-

• Hydrogen vessel with glass windows was employed; then, crack initiation behavior of the rubber during pressurization was also observed.



Camera: SONY a900 DSLR-A900, SAL100M28



Results and discussion-10 MPa hydrogen exposure-

- The specimen was exposed to hydrogen gas at 10 MPa and 25 °C for 65 h.
- No cracks were observed during hydrogen exposure and during decompression.
- Many cracks were observed after decompression, and grew with the elapsed time after decompression.



Results and discussion-2 MPa hydrogen exposure-

- The specimen was exposed to hydrogen gas at 2 MPa and 25 °C for 65 h.
- Cracks initiated after decompression as well as 10 MPa.
- The crack damage at 2 MPa was slighter that that at 10 MPa



Results and discussion-Relationship between hydrogen pressure and internal pressure-

The internal pressure (Π) of a bubble is nearly equal to hydrogen pressure (p).
 When a hydrogen pressure becomes higher, the internal pressure of the bubble also becomes higher.





Results and discussion-Influence of hydrogen pressure on crack damage-

- The crack damage was more serious with an increase in hydrogen pressure.
- The hydrogen pressure at crack initiation ranged from 1.5 MPa to 2.0 MPa.





Results and discussion-3 MPa hydrogen exposure-

·Micrometer-size cracks initiated from the interior of rubber material.



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Results and discussion-Relationship between crack length and time-

 The cracks grow with the elapsed time after decompression. This behavior is related to crack growth under constant loading.
 This type of crack growth is generally referred to as static crack growth.



Elapsed time after decon



Results and discussion-Static crack growth behavior-

- Constant strain was applied to cut specimen.
- Relationship between static crack growth rate and tearing energy is obtained.





Results and discussion-SEM images of fracture surfaces-

Fracture origins can be divided into two types.

- : Fracture from the site with micrometer-size defect. Case I
- : Fracture from the site without anything. Case II



Case I





Results and discussion-Suggested fracture processes of internal fracture-



nanoscale fracture process by optical microscopy.



Bubble formation by change in free energy





Results and discussion-Detection of Nanoscale fracture by AE and AFM-





Introduction-Acoustic emission (AE)-

- ・アコースティックエミッション(AE)とは き裂進展や塑性変形によって材料内部で発生する弾性波現象 (主として20kHz以上の弾性波を対象)
- ・AE法

AEを検出して,内部き裂を非破壊で検知する方法であり,金属 材料,コンクリートおよびCFRPのき裂検知に使用されている.





Experimental-AE measurement in air-

- ・室温(20~25 ℃)・大気中において,引張試験および静き裂進展試験を 実施し,発生するAEを計測.
- ・水素曝露した試験片では,減圧後5~10分から室温・大気中において AE計測を開始.

Room temperature in air





Results and discussion-Preliminary tests by AE-

- As preliminary tests, tensile test and static crack growth test were conducted.
- AE signals were detected at crack growth test, whereas these were hardly detected at tensile test.

This result implies that the AE signals were mainly generated during fracture process of chain molecules, not deformation process.





Results and discussion-Hydrogen exposure tests by AE-

- AE event counts and amplitude increased with an increase in hydrogen pressure.
- In spite of the specimen exposed to 0.6 MPa hydrogen gas where no cracks initiate, AE signals were detected.

Nanoscale fracture occurs although no cracks are observed by optical microscopy. This result implies sub-micromter-size bubbles initiate although no cracks initiate.



Experimental-Preparation of specimen for AFM-



25

Results and discussion-AFM observation of rubber structure-

- In spite of unexposed specimen, nanoscale dents were observed. The rubber structure is not homogeneous at nanometer size.
- The length and number of the nanoscale dents increased by hydrogen exposure. Nanoscale fracture occurs by hydrogen exposure.





Results and discussion-AFM observation of rubber structure-

• When the scatter of data is considered, the number of nanoscale dents increased by hydrogen exposure.

Number of observation areas: 10 Observation area: $2 \ \mu m \times 2 \ \mu m$





Results and discussion-suggested nanoscale rubber structure and fracture-





Results and discussion-Summary of multi-scale fracture processes-

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Results and discussion-Estimation of critical hydrogen pressure-



Results and discussion-Estimation of critical hydrogen pressures-

- Bubble formation
 - $T \geq \Gamma_{\rm BF}$

It is considered that the values of $\Gamma_{\rm BF}$ are not constant.

Therefore, the minimum value is assumed to be 50×10^{-3} N/m. (This value corresponds with the energy when bubbles grow

in simple liquids.)

Crack initiation

$$T \ge \Gamma_{\rm CI} = T_{\rm s,th} (= 55 \,{\rm N/m})$$



Items	Experiment (MPa) ^{*1}	Method	Energy (N/m)	Size (µm)	Estimation (MPa) ^{*1}
Bubble formation (Nanoscale fracture)	0.5–0.6	Acoustic emission	50 × 10 ⁻³	0.3	0.57
Crack initiation (Microscale fracture)	1.5–2.0	Optical microscope	55 ^{*2}	14	1.53

*1) Gage pressure (=absolute pressure -0.1 MPa)

*2) This value was obtained from static crack growth tests.



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